## LOYOLA COLLEGE (AUTONOMOUS), CHENNAI - 600 034



## M.Sc. DEGREE EXAMINATION - CHEMISTRY

## SECOND SEMESTER - APRIL 2024

## PCH2MC02 - COORDINATION CHEMISTRY

	Date: 06-04-2024 Dept. No. Time: 01:00 PM - 04:00 PM	Max.: 100 Marks
	SECTION A – K1 (CO1)	
		(5 1 5)
	Answer ALL the questions $(5 \times 1 = 5)$	
1	Answer the following.	
a)	Differentiate the absorption spectra of complexes with and without Jahn Teller distortion.	
b)	Give an example for complementary reaction.	
c)		
d)	Differentiate between ENDOR and ELDOR.	
e)	Mention any two functions of metals in metalloproteins.	
	SECTION A – K2 (CO1)	
	Answer ALL the questions	$(5 \times 1 = 5)$
2	Fill in the blanks.	
a)	[Fe(H <sub>2</sub> O) <sub>6</sub> ] <sup>2+</sup> complex is having B.M. of spin only magnetic moment.	
b)	Isotactic polypropene is prepared by using	
c)	A Tanabe-Sugano diagram is obtained by plotting	
d)	Only one line found in the halogen NQR spectrum of K <sub>2</sub> PtCl <sub>6</sub> is consistent with structure.	
e)	Ferritins play a key role in	
	SECTION B – K3 (CO2)	
	Answer any THREE of the following	$(3 \times 10 = 30)$
3	Explain the application of crystal field theory in ionic radii, hydration enthalpy and lattice energy.	
4	Deduce the mechanism of photorearrangement reactions with an example.	
5	How is IR spectroscopy useful for the detection of a) terminal and bridging car	rbonyls, b) linkage
	isomers of NO <sub>2</sub> and CN <sup>-</sup> ?	
6	How are Mossbauer spectra useful in determining the oxidation state of atoms?	
7	Explain the structure, mode of action and functions of Horseradish Peroxidase.	
	SECTION C – K4 (CO3)	
	Answer any TWO of the following	$(2 \times 12.5 = 25)$
8	What is meant by LCAO? Draw and write the wave functions of various orbita	als involved in LCAO
	method employed to explain the bonding in coordination complexes.	
9	Discuss the mechanism of inner sphere mechanism of electron transfer reactions.	

10	With a suitable example, explain the electronic spectrum of d <sup>3</sup> and d <sup>7</sup> complexes using Orgel		
	diagram.		
11	Discuss Mossbauer spectrum of any three coordination complexes.		
	SECTION D – K5 (CO4)		
	Answer any ONE of the following $(1 \times 15 = 15)$		
12	a) Represent schematically the d orbital splitting in the case of d <sup>6</sup> system considering both low spin and high spin octahedral complexes.		
	b) How does valence bond theory determine hybridization, geometry and magnetic nature of the		
	coordination complex? Give a suitable example. $(6+9)$		
13	Interpret the structure and functions of hemoglobin.		
	SECTION E – K6 (CO5)		
	Answer any ONE of the following $(1 \times 20 = 20)$		
14	How are the following metal complexes investigated using ESR experiments?		
	(i) $[Mo(CN)_8]^{3-}$ (ii) bis(salicylaldimine)Cu(II) complex. (10 + 10)		
15	a) Write a note on the electronic spectra of lanthanide complexes.		
	b) Explain the biological functions of azurin and stellacyanin. $(10 + 10)$		

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